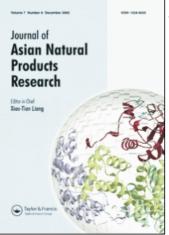
This article was downloaded by: On: 22 January 2011 Access details: Access Details: Free Access Publisher Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713454007

Two New Taxane Diterpenoids from the Seeds of the Chinese Yew, Taxus

yunnanensis

Qing-Wen Shi^a; Takayuki Oritani; Takeyoshi Sugiyama^a; Ryo Murakami^a; Teiko Yamada^a ^a Laboratory of Applied Bioorganic Chemistry, Division of Life Sciences, Graduate School of Agricultural Science, Tohoku University, Aoba-ku, Sendai, Japan

To cite this Article Shi, Qing-Wen , Oritani, Takayuki , Sugiyama, Takeyoshi , Murakami, Ryo and Yamada, Teiko(1999) 'Two New Taxane Diterpenoids from the Seeds of the Chinese Yew, *Taxus yunnanensis*', Journal of Asian Natural Products Research, 2: 1, 71 - 79

To link to this Article: DOI: 10.1080/10286029908039894 URL: http://dx.doi.org/10.1080/10286029908039894

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

JANPR, Vol. 2, pp. 71-79 Reprints available directly from the publisher Photocopying permitted by license only () 1999 OPA (Overseas Publishers Association) N.V. Published by license under the Harwood Academic Publishers imprint, part of The Gordon and Breach Publishing Group. Printed in Malaysia.

TWO NEW TAXANE DITERPENOIDS FROM THE SEEDS OF THE CHINESE YEW, TAXUS YUNNANENSIS

QING-WEN SHI, TAKAYUKI ORITANI*, TAKEYOSHI SUGIYAMA, RYO MURAKAMI and TEIKO YAMADA

Laboratory of Applied Bioorganic Chemistry, Division of Life Sciences, Graduate School of Agricultural Science, Tohoku University, 1-1 Tsutsumi-dori Amamiya, Aoba-ku, Sendai 981-8555, Japan

(Received 9 March 1999; Revised 21 May 1999; In final form 15 June 1999)

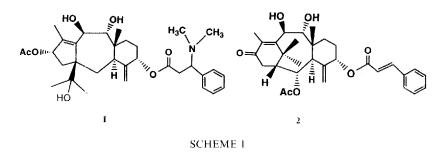
A new taxoid and a taxinine analogue were isolated from the seeds of the Chinese yew, *Taxus yunnanensis*. The structures were established as 13α -acetoxy- 5α -(3'-dimethylamino-3'-phenyl)-propionyloxy-11($15 \rightarrow 1$)-*abeo*taxa-4(20),11-diene- 9α ,10 β -diol and 2α -acetoxy- 5α -cinnamoyloxy- 9α ,10 β -dihydroxy-taxa-4(20),11-diene-13-one on the basis of 1D, 2D NMR, and MS spectral analysis.

Keywords: Taxus yunnanensis; Taxaceae; Taxoids; $11(15 \rightarrow 1)$ -Abeotaxane; Seeds

INTRODUCTION

The genus *Taxus* comprises about 10 species in the world, five of them are distributed in China and southeastern Asia. *Taxus yunnanensis*, which was indigenous to China, is an evergreen tall tree or shrub mainly distributed in the wet valley area of Yunnan, Sichuan Provinces, and Tibet Autonomous Region in the south-west of China. Previous studies on taxane diterpenes in the bark, leaves, branches, and root of this plant have resulted in the isolation of more than 40 new taxane diterpenoids [1–22]. In the course of our studies on the yew tree, we have investigated the constitutions of the leaves, bark, and seeds of Chinese yew, *Taxus chinensis* var. *mairei* [23–27]. In view

^{*} Corresponding author. Tel./Fax: 081-22-717 8783. E-mail: shi@biochem.tohoku.ac.jp.



of no report on the constitutions of seeds of T. *yunnanensis*, recently we investigated the component of the seeds of this plant, and led to the isolation of two new taxane diterpenoids (Scheme 1). In this communication we describe the isolation and characterization of two compounds.

RESULTS AND DISCUSSION

A methanolic extract of the seeds of T. yunnanensis was processed as described in the Materials and Methods section to afford two new taxane diterpenoids (1 and 2). Compound 1 was isolated as a colorless gummy substance in a yield of 0.002% on the dry material. EI-MS produced an ion peak at m/z 569 ([M]⁺). The molecular formula of compound 1, C₃₃H₄₇O₇N, was deduced from combined analysis of HR-EI-MS at m/z 569.3353 ([M]⁺) and ¹³C-NMR spectrum. Intensive absorptions at 3400 and 1730 cm⁻¹ in the IR spectrum implied that 1 possesses hydroxyl and ester groups, respectively. The ¹H-NMR spectrum of 1, tabulated in Table I, exhibited the proton signals due to the four methyl groups at δ 0.83, 1.15, 1.29, and 1.92 ppm, which were the characteristic signals of the taxane skeleton. One acetyl group resonated at a relatively lower field (δ 2.15 ppm), which was verified by observation of ¹³C-NMR signals at δ 171.03 and 20.97 ppm. These signals suggested that 1 had a taxane-type skeleton. The connectivities of the protons at the taxane skeleton of 1 were determined by analysis of the ¹H-¹H COSY spectrum. Interpretation of ¹H-, ¹³C-NMR and HMBC spectra permitted the positional assignment of functional groups. The ¹H-NMR signals at δ 5.07 (1H, br s), 4.67 (1H, br s) and 2.75 (1H, d, J = 7.70 Hz) are characteristic of an exocyclic methylene and C-3 ring junction proton in a taxa-4(20),11-diene, respectively [28]. Additionally, four oxygen-bearing one-proton signals appeared at lower field. Of them, the signal at δ 3.97 ppm (1H, d, J = 9.61 Hz), which showed cross-peaks with

	IABLET	I- and C-INMIX spe		
Position	^{1}H	J	$^{1}H^{-1}HCOSY$	¹³ C
1				62.29
2a	1.88 m		H-3 α	28.81
2b	1.28 m		H-2 α	
3	2.75 d	7.70	H-2 α	40.11
4				147.71
5	5.22 br s		H-6	75.16
6	1.60 m		H-7	27.43
7	1.38 m		H-6	25.03
8				41.26
9	3.97 d	9.61	H-10 α	79.39
10	4.40 br d	9.61	H-9 <i>β</i>	69.41
11				140.82
12				142.03
13	5.51 t	7.14	H-14, 18-CH3	80.07
14α	1.13 m		H-14β	44.07
14β	2.36 dd	6.47, 13.24	H-14 α , 13 β	
15				76,50
16	1.29 s		H-17	29.71
17	1.15 s		H-16	26.59
18	1.92 br s		H-13	11.35
19	0.83 s		••	17.19
20a	5.07 br s		H-20b	111.46
20Б	4.67 br s		H-20a	
13-OAc	2.15 s			20.97, 171.03
1'				170.66
2'	2.88 dd	6.97, 13.74	2". 3'	39.69
2"	2.66 dd	9.07, 13.74	2' 3'	0,10,
3'	3.77 t	8.30	2", 3' 2', 3' 2', 2"	67.13
4'		0.00	-,-	138.57
5'	7.29 m			128.27
6'	7.29 m			128.12
7'	7.29 m			120.12
N-CH ₃	2.17 s			42.45

TABLE I ¹H- and ¹³C-NMR spectral data of 1 in CDCl₃

19-CH₃, C-11 in the HMBC spectrum, was attributed to H-9 β . The signal at δ 4.40 ppm (1H, br d, J = 9.61 Hz), which showed a cross peak with H-9 α in the ¹H-¹H COSY experiment, and showed cross-peaks with C-11 and C-12 in the HMBC spectrum, was assigned to H-10 α . Large vicinal coupling indicated a *trans*-oriented configuration of the H-9 β and H-10 α . The spin system derived from 18-CH₃, H-13 β , H-14 α , and H-14 β was readily interpreted. The signal of three protons as a doublet at δ 1.92 ppm was assigned to 18-CH₃ based on the long-range coupling with H-13 β ; the triplet at δ 5.51 ppm (1H, t, J = 7.14 Hz), was assigned to H-13 β ; the doublet of doublets at 2.36 ppm and the multiplet at δ 1.13 ppm were assigned to the C-14 methylene protons, H-14 β and H-14 α , respectively, based on their geminal coupling and coupling to H-13 β . The signal at δ 5.22 ppm (1H, br s) was

characteristic signal of H-5 β . All of the proton-bearing carbons were assigned by an analysis of the HETCOR spectrum. Four oxygen-containing carbons (C-5, C-9, C-10, and C-13) were correlated with their corresponding proton signals. The H-10 α signal showed cross-peaks with the resonances at δ 140.82, 142.03 and 62.29 ppm, which were assigned for C-11, C-12, and C-1. respectively. The C-11 and C-12 carbon signals showed cross-peaks with the H-14 β resonance, indicated that both C-11 and C-12 are three bonds apart from H-14 β . This means that the A ring was a cyclopentene as in an $11(15 \rightarrow 1)$ abeotaxane structure [29,30]. The carbon signal at δ 76.50 ppm, assigned to the hydroxyl-bearing C-15, displayed a cross-peak with the C-16 and C-17 methyl resonances at δ 1.29 and 1.15 ppm. The C-1 signal (δ 62.29 ppm), apart from H-10 α , also showed three-bond coupling with the H-3 α and C-16, C-17 methyl signals. Since no cross-peak was observed between C-16. C-17 (methyl) signals and the C-11 olefinic carbon in the HMBC spectrum further supported the $11(15 \rightarrow 1)abeotaxane$ skeleton for 1 [31]. The presence of a Winterstein acid [(3'-dimethylamino-3'-phenyl)-propionyloxyl] moiety in 1 was suggested from the signals at δ 2.17 (6H. s). 2.88 (1H, dd, J = 13.74, 6.97 Hz). 2.66 (1H, dd, J = 13.74, 9.07 Hz). 3.77 (1H, t, J = 8.30 Hz), and 7.29 ppm in the ¹H-NMR spectrum, and the signals at 8 170.66, 39.69, 67.13, 138.57, 128.27, 128.12, 127.67, and 42.45 ppm in the ¹³C-NMR spectrum, in good agreement with literature values [32]. Further support was provided by the fragment ions in the EI-MS at m/z 192 and 134 (base-peaks), which were the characteristic fragments of Winterstein acid [33-35]. The location of the Winterstein acid moiety was deduced at C-5 from the HMBC spectrum (Fig. 1). The relative stereochemistry of the terpenoid skeleton of 1 was determined from chemical shifts, coupling constants and NOESY experiment. A coupling constant between H-9 and H-10 of J = 9.61 Hz indicated that the B-ring was the chair boat conformation [29]. The NOESY experiment established the

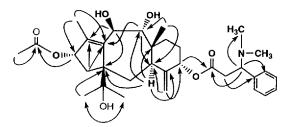


FIGURE 1 Long-range $H \rightarrow C$ correlations observed by the HMBC spectrum for 1.

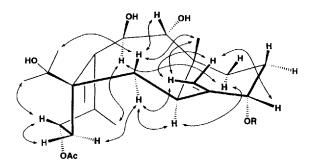


FIGURE 2 Relative stereochemistry of 1, proposed by NOESY experiment (600 MHz).

relative stereochemistry of 1 at all the position, and the results are shown in Fig. 2. Thus the structure of 1 was determined as 13α -acetoxy- 5α -(3'-dime-thylamino-3'-phenyl)-propionyloxy- $11(15 \rightarrow 1)$ -abeotaxa-4(20),11-diene- 9α , 10β -diol.

Compound 2, which was isolated as a colorless gummy substance, had a molecular ion peak at m/z 522 in its EI-MS (HR-EI-MS, C₃₁H₃₈O₇); the base peak was at m/z 131 due to the cinnamoyl ion (C₉H₇O), and a peak at m/z 374 arose from the loss of cinnamic acid (C₉H₈O₂) from M⁺. The IR spectrum had bands at 3450 (hydroxy), 3050 (aromatic), 1730 (ester), 1710 (conjugated C=O), 1660, 1620 (benzene), and 1625 (olefinic) cm⁻¹. The UV spectrum had λ_{max} at 279 nm due to the cinnamate ester group. The ¹H-NMR spectrum had well-dispersed signals suggestive of a taxane derivative containing one acetate group and one cinnamate group. The fact that 18-CH₃ was a sharp singlet suggested that C-13 did not have a hydrogen attached, but instead bore a ketone function. In accordance with this, H₂-14 displayed a large $J_{gem} = 19.78$ Hz. By means of the ${}^{1}H - {}^{1}H$ COSY spectrum, the complete connection network was established for H-14 α -H-14 β -H- 1β -H-2 β -H-3 α -H-20a-H-20b-H-5 β H₂-6-H-7 α . H-3 α and H-5 β had allylic coupling with H-20a and H-20b. The lower-field chemical shift of H- 2α indicated an acetoxy group located at C-2. A pair of distinguished doublcts with a large coupling constant at δ 4.89 and 4.19 ppm were attributed to H-10 α and H-9 β , respectively, their chemical shifts indicated there are hydroxyl groups attached to C-9 and C-10, i.e. which was 9,10-dideacetyl taxinine. Its ¹H-NMR spectrum was closely comparable with that of taxinine with the exception of that H-9 β and H-10 α were shifted upfield. The cinnamate group was located at C-5 α in accordance with what has been observed in the other taxinine derivatives [28,36-38]. Therefore, the structure of **2** was elucidated as 2α -acetoxy- 5α -cinnamoyloxy- 9α ,10 β -dihydroxy-taxa-4(20),11-diene-13-one. Compound **2** can be formed chemically from taxinine [39], but this is the first report as a natural product.

EXPERIMENTAL SECTION

General Experimental Procedures

Optical rotations were recorded on a Horiba SEPA-300 digital polarimeter. UV spectrum was run on a Shimadzu UV-1600 spectro-photometer. IR spectra were obtained on a Jasco IR-810 instrument. MS were measured on a Jeol JMS-700 spectrometer using EI modes. ¹H- and ¹³C-NMR spectra were obtained Varian Unity Inova 600 spectrometers operating at 600 MHz for ¹H. 150 MHz for ¹³C nucleus, in CDCl₃ at 20°C, chemical shifts are expressed in parts per million scale relative to that of tetramethylsilane (TMS, $\delta = 0$) as an internal standard, and coupling constants are given in Hertz. ¹H-¹³C HETCOR and HMBC experiments were performed on the same spectrometer, using standard Varian pulse sequences. ¹H-¹H COSY spectrum was measured on Varian GEMINI 2000/300 spectrometer at 300 MHz. Open column chromatography (CC) was performed using Merck silica gel 60 (100- 200 mesh). Thin layer chromatography (TLC) was carried out with the precoated Merck silica gel 60 F_{254} plates. Preparative TLC were performed using the same type of plates as used for TLC but with 0.85 mm (dried for 24 h at room temperature and activated for 4 h at 120°C) thickness, the spots were detected under UV (254 nm) and/or by spraying with 10% sulfuric acid and then heating on a hot plate.

Plant Material

The seeds of *T. yunnanensis* were collected in Congteng county, Yunnan Province, in the south-west of China, in October of 1995. The botanical identification was made by Prof. J.H. Wang, School of Pharmaceutical Science, Hebei Medical University, the People's Republic of China. A voucher specimen has been deposited in our laboratory of Graduate School of Agricultural Science, Tohoku University, Japan.

Extraction and Isolation

Air dried seeds (2.2 kg) were crushed and extracted with hexane three times at room temperature to remove major part of nondesired neutral component.

The residue was extracted three times with methanol (MeOH), the MeOH extracts were condensed to residue (135 g) under reduced pressure. Subsequently this residue was diluted with water and was extracted five times with EtOAc (85 g). The combined EtOAc layer was further extracted with 5% HCl. After neutralization, the aqueous layer was extracted three times with EtOAc. The combined EtOAc extract, upon evaporation, yielded 8.8 g of yellowish syrup, which was subjected to CC, eluted with hexane–ethyl acetate (2:1, 1:1, 1:2, 1:4), 12 fractions were obtained, and fraction 3 (900 mg) was further separated by preparative TLC repeatedly with different developing solvent (CHCl₃–MeOH, 100:4.5; hexane–EtOAc, 1:2; hexane–acetone 5:3), and finally compound **1** (2.5 mg) and **2** (1 mg) were separated.

13α-Acetoxy-5α-(3'-dimethylamino-3'-phenyl)-propionyloxy-11(15 \rightarrow 1)abeotaxa-4(20),11-diene-9α,10β-diol (1) Gum, $[\alpha]_D^{24} - 27$ (c 0.01, CHCl₃). IR (film, CHCl₃) ν_{max} : 3400, 2940, 2850, 2850, 2780, 1730, 1650, 1450, 1430, 1365, 1240, 1180, 1020, 750 and 700 cm⁻¹; El-MS: *m/z* (rel. int.): 569 ([M]⁺) (19), 509 ([M-AcOH]⁺) (8), 451 (37), 192 ([HOCOCH₂CH(Me₂N)Ph]⁺) (76), 134 ([Me₂N=CHPh]⁺) (100), 105 (10), and 43 (9). HR-EI-MS: 569.3353 (calcd. for C₃₃H₄₇O₇N, 569.3350), The ¹H- and ¹³C-NMR spectral data see Table I.

 2α -Acetoxy- 5α -cinnamoyloxy- 9α , 10β -dihydroxy-taxa-4(20), 11-diene-13one (2) Colorless gum; $[\alpha]_{D}^{25}$ +132 (c 0.01, CHCl₃); UV (MeOH) λ_{max} $(\log \epsilon)$ 279 (4.3) nm; IR (film, CHCl₃) ν_{max} : 3450, 3050, 2925, 1730, 1710, 1660, 1625, 1620, 1440, 1370, 1310, 1240, 1170, 1030, and 750 cm⁻¹; EI-MS m/z (rel. int.): 522 (30) ([M]⁺), 374 (18) ([M-cinnamic acid]⁺), 131 (100), 103 (98) 91 (100), 77 (65), and 43 (100). HR-EI-MS: m/z 522.2621 (calcd. for C₃₁H₃₈O₇, 522.2615); ¹H-NMR (ppm, 300 MHz): δ 2.14 (1H, m, H-1), 5.52 (1H, br d, J = 6.32 Hz, H-2), 3.36 (1H, d, J = 6.32 Hz, H-3), 5.32 (1H, br s, H-5), 1.95 (2H, m, H-6), 1.75 (2H, m, H-7), 4.19 (1H, d, J=9.6 Hz, H-9), 4.89 (1H, br d, J = 9.6 Hz, H-10), 2.42 (1H, br d, J = 19.78 Hz, H-14 α), 2.83 $(1H, dd, J = 19.78, 7.42 Hz, H-14\beta)$, 5.32 (1H, br s, H-20a), 4.84 (1H, br s, H-20b), 6.44 (1H, d, J = 15.93 Hz, H-2'), 7.65 (1H, d, J = 15.93 Hz, H-3'), 7.76 (2H, d, J = 6.87 Hz, H-5',9'), 7.42 (3H, m, H-6',7',8'), 1.22 (3H, s, 16-CH₃), 1.71 (3H, s, 17-CH₃), 2.12 (3H, s, 18-CH₃), 1.10 (3H, s, 19-CH₃), 2.16 (3H, s, CH₃CO-). ¹³C-NMR (75 MHz, CDCl₃): 48.82 (C-1), 69.73 (C-2), 43.17 (C-3), 142.58 (C-4), 78.74 (C-5), 28.44 (C-6), 26.16 (C-7), 44.41 (C-8), 77.76 (C-9), 73.41 (C-10), 155.37 (C-11), 135.86 (C-12), 200.05 (C-13), 36.05 (C-14), 37.54 (C-15), 25.14 (C-16), 37.91 (C-17), 14.06 (C-18), 17.58 (C-19), 118.08 (C-20), 169.87 (CH₃CO-), 166.53 (C-1'), 21.30(CH₃CO-).

Acknowledgments

The authors are grateful to Mrs. Yuhko Sugiyama for measuring the NMR spectra. The financial support for the work described here comes from the Ministry of Education, Science, Sports, and Culture of Japan through a grant-in-aid for scientific research.

References

- [1] W.M. Chen, P.J. Zhang and Q.T. Zheng, Acta Pharm. Sin. 1991, 26, 747-754.
- [2] Z.P. Zhang and Z.J. Jia. Phytochemistry 1990, 29, 3673 3675.
- [3] W.M. Chen, J.Y. Zhou, P.L. Zhang and Q.C. Fang. Chin. Chem. Lett. 1993, 4, 699-702.
- [4] C. Rao, J.Y. Zhou, W.M. Chen, Y. Lu and Q.T. Zheng, Chin. Chem. Lett. 1993, 4. 693 694.
- [5] W.M. Chen, J.Y. Zhou, P.L. Zhang and Q.C. Fang. Chin. Chem. Lett. 1993, 4, 695–698.
- [6] W.M. Chen, P.L. Zhang, B. Wu and Q.T. Zheng. Chin. Chem. Lett. 1991. 2, 441 442.
- [7] Q. Yue, Q.C. Fang, X.T. Liang, C.H. He and X.L. Jing, Planta Med. 1995, 61, 375 377
- [8] Q. Yue, Q.C. Fang and X.T. Liang. *Phytochemistry* 1996, 43, 639 642.
 [9] Q. Yue, Q.C. Fang, X.T. Liang and C.H. He. *Phytochemistry* 1995, 39, 871 873.
- [10] W.M. Chen, P.L. Zhang and J.Y. Zhou. Acta Pharm. Sin. 1994, 29, 207 214.
- [11] Z.Y. Chen, C.W. Gao and Y.S. Chen. Acta Botanica Sin. 1996, 38, 323-327.
- [12] X.K. Liu, D.G. Wu and Z.Y. Wang. Kexue Tongbao 1992, 37, 2186-2189.
- [13] C. Rao, J.Y. Zhou, W.M. Chen, Y. Lu and Q.T. Zheng. Acta Pharm. Sin. 1994. 29. 355 359.
- [14] W.M. Chen, P.L. Zhang, J.Y. Zhou, X. Liu and Q.C. Fang. Acta Pharm. Sin. 1994. 29. 751 757.
- [15] J.Y. Zhou, P.L. Zhang, W.M. Chen and Q.C. Fang. Phytochemistry 1998, 48. 1387 1389.
- [16] S.Z. Zhong, Z.X. Hua and J.S. Fan. J. Nat. Prod. 1996, 59, 603–605.
- [17] S.X. Zhang, C.T. Lee, Y. Kashiwada, K. Chen, D.C. Zhang and K.H. Lee, J. Nat. 1994. 57, 1580 1583
- [18] H.J. Zhang, T. Yoshio, M. Takashi, M. Yoshinori, Y. Kenichiro, W. Xiang, Q. Mu and H.D. Sun. Heterocycles 1994, 38, 975--980.
- [19] H.J. Zhang, T. Yoshio, M. Takashi, M. Yoshinori, Y. Kenichiro, W. Xiang, Q. Mu and H.D. Sun, Chin, Chem, Lett. 1994, 5, 957–960.
- [20] H.J. Zhang, Y. Takeda and H.D. Sun. Phytochemistry 1995, 39, 1147-1151.
- [21] H.J. Zhang, H.D. Sun and Y. Takeda. Chin. Chem. Lett. 1995, 6, 479–482.
- [22] H.J. Zhang, Q. Mu, W. Xiang, P. Yao, H.D. Sun and Y. Takeda. Phytochemistry 1997, 44. 911 915
- [23] Q.W. Shi, T. Oritani, H. Kiyota and T. Horiguchi. Nat. Prod. Lett. 1998, 12, 67-74.
- [24] Q.W. Shi, T. Oritani, T. Sugiyama and H. Kiyota. Planta Med. 1999, 64, 766-769.
- [25] Q.W. Shi, T. Oritani, T. Sugiyama and H.J. Kiyota. Nat. Prod. 1998, 61, 1437–1440.
- [26] Q.W. Shi, T. Oritani and T. Sugiyama. Phytochemistry 1999. 50, 633-636.
- [27] Q.W. Shi, T. Oritani, T. Sugiyama, H. Kiyota and T. Horiguchi. Heterocycles 1999, 51. 841 850.
- [28] G. Appendino. In The Chemistry and Pharmacology of Taxol and its Derivatives. Ed Farina, V., Amsterdam, 1995, Vol. 22, pp. 55-101.
- [29] K. Fuji, K. Tanaka, B. Li, T. Shingu, T. Yokoi, H.D. Sun and T. Taga. Tetrahedron 1995. 51. 10 175--10 188.
- [30] S. Zhang, C.T. Lee, T. Che, Y. Kashiwad, D. Zhe, A. McPhail and K.J. Lee, Chem. Soc. Chem. Commun. 1994, 1561-1563.
- [31] G. Appendino, L. Barboni, P. Gariboldi, E. Bombardelli, B. Gabetta and D.J. Viterbo Chem. Soc., Chem. Commun. 1993, 1587–1588.

- [32] G. Appendino, H.C. Ozen, I. Fenoglio, P. Garboldi, B. Gabetta and E. Bombardelli. *Phytochemistry* 1993, 33, 1521–1523.
- [33] R.P. Doss, J.R. Carney, C.H. Shanks, R.T. Williamson and J.D. Chamberlain. J. Nat. Prod. 1997, 60, 1130–1133.
- [34] J.Z. Zhang, Q.C. Fang, X.T. Liang, C.H. He, M. Kong, W.Y. He and X.L. Jin. Phytochemistry 1995, 40, 881-884.
- [35] J.Z. Zhang, Q.C. Fang, X.T. Liang and C.H. He. Chin. Chem. Lett. 1994, 5, 497-501.
- [36] G. Appendino, G. Cravotto, R. Enriu, D. Gariboldi, L. Barboni, E. Torregini, G. Gabetta, G. Zml and E. Bombardelli. J. Nat. Prod. 1994, 57, 607–613.
- [37] D.G.I. Kingston, A.A. Molinero and J.M. Rimoldi. In Progress in the Chemistry of Organic Natural Products. Eds. Herz, W., Kirby, G.W., Moore, R.E., Steglich, W. and Tamm, C.H., Springer, 1993, Vol. 61, pp. 1–206.
- [38] G. Appendino. Nat. Prod. Rep. 1995, 12, 349-360.
- [39] M. Sako, H. Suzuki, N. Yamamoto, K. Hirota and Y. Maki, J. Chem. Soc., Perkin Trans. 1, 1998, 714-421.